This article was downloaded by: On: 22 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713454007

Two triterpenes from Lysimachia foenum-graecum

Xiang-Ri Li^a; Zhi-Meng Li^b; Shu-Shan Du^c; Gang-Li Wang^d; Rui-Chao Lin^d ^a School of Chinese Pharmacy, Beijing University of Traditional Chinese Medicine, Beijing, China ^b Scientific Research Institute of Beijing Tongrentang Pharmaceutical Company, Beijing, China ^c College of Resource Science & Technology, Center for Natural Medicine Engineering, Ministry of Education, Beijing Normal University, Beijing, China ^d National Institute for the Control of Pharmaceutical and Biological Products, Beijing, China

To cite this Article Li, Xiang-Ri , Li, Zhi-Meng , Du, Shu-Shan , Wang, Gang-Li and Lin, Rui-Chao(2009) 'Two triterpenes from *Lysimachia foenum-graecum*', Journal of Asian Natural Products Research, 11: 2, 128 — 131 **To link to this Article: DOI:** 10.1080/10286020802573859

URL: http://dx.doi.org/10.1080/10286020802573859

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Two triterpenes from Lysimachia foenum-graecum

Xiang-Ri Li^a*, Zhi-Meng Li^b, Shu-Shan Du^c, Gang-Li Wang^d and Rui-Chao Lin^d

^aSchool of Chinese Pharmacy, Beijing University of Traditional Chinese Medicine, Beijing, China; ^bScientific Research Institute of Beijing Tongrentang Pharmaceutical Company, Beijing, China; ^cCollege of Resource Science & Technology, Center for Natural Medicine Engineering, Ministry of Education, Beijing Normal University, Beijing, China; ^dNational Institute for the Control of Pharmaceutical and Biological Products, Beijing, China

(Received 23 May 2008; final version received 27 September 2008)

One new oleane-type triterpene saponin, named lysimachiagenoside A (1) and the known 21-O-angeloylbarringtogenol C (2) were isolated from the aerial parts of Lysimachia foenumgraecum Hance. 21-O-angeloylbarringtogenol C was a new natural product. These structures were identified on the basis of 1D- and 2D-NMR techniques, including ${}^{1}H^{-1}H$ COSY, HMQC, HMBC, TOCSY, and ROESY experiments as well as chemical methods.

Keywords: Lysimachia foenum-graecum Hance; triterpene saponin; lysimachiagenoside A

Introduction 1.

Lysimachia foenum-graecum Hance (Primulaceae), distributed mainly in Guangxi and Yunnan Provinces of China, has been commonly used as perfume plant and pest repellent. In Chinese folk medicine, the plant has also been used for the treatment of cold and headache [1]. It has been reported that some plants of the genus Lysimachia contained saponins and flavones [2-6]. In our recent study, lysimachiagenoside A and 21-O-angeloylbarringtogenol C were first isolated from the aerial parts of L. foenumgraecum. Lysimachiagenoside A was a new oleanane-type triterpenoid saponin and 21-Oangeloylbarringtogenol C was a new natural product. Their structures were identified by 1D- and 2D-NMR techniques. In this paper, we described the isolation and structural elucidation of these two triterpenoid saponins.

Results and discussion 2

Compound 1 was obtained as a white powder. The ESI-MS of 1 showed a pseudo-molecular ion $[M+Na]^+$ at m/z1115.6, compatible with the molecular formula C53H88O23, which was further determined by HR-FAB-MS at m/z1115.5574 [M+Na]⁺. Briefly, the analysis of the NMR spectral data indicated that 1 was a saponin consisting of a triterpene aglycone and four monosaccharides. The ¹³C NMR spectrum of 1 showed 53 carbon signals, from which 23 were assigned to four monosaccharide units and 30 to triterpene aglycone moiety (Table 1). Detailed comparison of the ¹³C and ¹H NMR spectral data of 1 with those reported in the literature suggested that the aglycone of 1 was barringtogenol C [7]. Four monosaccharide units were determined from the TOCSY spectrum with the aid of COSY, HMOC, and

ISSN 1028-6020 print/ISSN 1477-2213 online © 2009 Taylor & Francis DOI: 10.1080/10286020802573859 http://www.informaworld.com

^{*}Corresponding author. Email: lixiangri@sina.com

HMBC spectra. Starting from the anomeric proton signal at $\delta_{\rm H}$ 5.39 (1H, d, J = 7.6 Hz), six correlated carbon signals were observed in the TOCSY spectrum and determined in sequence to be at $\delta_{\rm C}$ 105.4 (C-1), 74.8 (C-2), 79.5 (C-3), 71.7 (C-4), 78.1 (C-5), and 62.8 (C-6), which suggested a glucosyl group. Similarly, another glycosyl group was identified from the analysis of the TOCSY spectra. A 6-deoxymonosaccharide, a rhamnosyl moiety, was elucidated from the methyl carbon at $\delta_{\rm C}$ 18.9 and the corresponding methyl proton at $\delta_{\rm H}$ 1.79 (1H, d, J = 5.8 Hz), together with the carbon signals at $\delta_{\rm C}$ 68.8 (C-5), 74.0 (C-4), and 72.7 (C-3). The remaining five carbon signals suggested the presence of a pentosyl group whose anomeric proton at $\delta_{\rm H}$ 4.94 (1H, brs) was only correlated to four carbon signals at $\delta_{\rm C}$ 103.5 (C-1), 80.8 (C-2), 72.1 (C-3), and 74.6 (C-4) in TOCSY spectrum, implying an arabinosyl group. The oxygen-bearing methylene at $\delta_{\rm C}$ 64.2 was assigned to be at the C-5 position of arabinosyl group based on ¹H-¹H COSY and HMBC experiments. The above inferences for the monosaccharide unit were further confirmed by TLC analysis of the acid hydrolysate of compound 1. The glycosidic position of the aglycone was determined to be at the C-3 position on the basis of HMBC correlation between the anomeric proton of arabinosyl group at $\delta_{\rm H}$ 4.94 (1H, brs) with C-3. The connectivity among the monosaccharide units was established with the following HMBC correlations: H-1 ($\delta_{\rm H}$ 5.39) of inner glucosyl group with C-2 ($\delta_{\rm C}$ 80.8) of arabinosyl group; H-1 ($\delta_{\rm H}$ 5.24) of outer glucosyl group with C-4 ($\delta_{\rm C}$ 74.6) of arabinosyl group; and H-1 ($\delta_{\rm H}$ 6.45) of rhamnosyl group with C-2 ($\delta_{\rm C}$ 77.4) of inner glucosyl group. The anomeric configurations of two glucosyl groups were determined to be β -orientated from the coupling constants of the anomeric protons. Similarly, the anomeric protons of rhamnosyl and arabinosyl groups were found to be in the α -orientation (Table 2). Thus, the complete structure of 1 was elucidated as barringtogenol C-3-O-αrhamnopyranosyl $(1 \rightarrow 2)$ - β -glucopyranosyl

Table 1. 13 C NMR spectral data of compounds 1 and 2 (500 Hz).

C No.	1^{a}	2 ^b	C No.	1	2
1	38.2	38.6	1′		169.9
2	25.7	27.0	2'		128.0
3	88.4	78.9	3'		138.5
4	39.4	38.7	4′		15.9
5	55.1	55.1	5'		20.5
6	17.8	18.3	Ara-1	103.5	
7	33.0	32.7	2	80.8	
8	39.9	39.7	3	72.1	
9	46.4	46.5	4	74.6	
10	36.7	36.9	5	64.2	
11	23.2	23.4	Glc'-1	102.5	
12	123.2	123.8	2	77.4	
13	135.6	141.5	3	76.6	
14	41.9	41.4	4	71.2	
15	33.7	33.7	5	78.0	
16	68.0	67.7	6	62.5	
17	48.1	46.9	Glc"-1	105.4	
18	40.5	40.9	2	74.8	
19	47.5	46.7	3	79.5	
20	36.4	35.3	4	71.7	
21	76.3	80.9	5	78.1	
22	76.5	78.6	6	62.8	
23	27.6	28.0	Rha-1	101.5	
24	16.0	15.6	2	72.3	
25	15.0	15.6	3	72.7	
26	16.3	16.7	4	74.0	
27	26.7	27.1	5	68.8	
28	68.2	66.5	6	18.9	
29	30.0	29.3			
30	20.2	19.8			

^a **1** in pyridine- d_5 .

^b**2** in DMSO.

Table 2. ¹H NMR spectral data for the sugar moiety of compound 1 (125 Hz, pyridine- d_5).

H No.	1	H No.	1
Ara-1	4.94 (brs)	Glc"-1	5.24 (d, 7.5)
2	4.55	2	4.17
3	4.69	3	4.16
4	4.26	4	4.13
5	4.32, 3.78	5	3.80
		6	4.45, 4.37
Glc'-1	5.39 (d, 7.6)	Rha-1	6.45 (brs)
2	4.04	2	4.51
3	4.29	3	4.59
4	4.16	4	4.53
5	4.09	5	5.00
6	4.35, 4.24	6	1.79 (d, 5.8)



Figure 1. Structures of compounds 1-2.

 $(1 \rightarrow 2)$ -[β -glucopyranosyl $(1 \rightarrow 4)$]- α -arabinopyranoside, named lysimachiagenoside A (Figure 1).

Compound **2** was obtained as a white powder. Its molecular formula $C_{35}H_{56}O_6$ was deduced from HR-FAB-MS at m/z 595.3979 $[M+Na]^+$. The structure of **2** (Figure 1) was determined to be 21-*O*-angeloylbarringtogenol C by the comparison with the reported spectral data [7], which was first found in natural products.

3. Experimental

3.1 General experimental procedures

Melting points were determined on a K micromelting point apparatus and are uncorrected. MS and HR-MS were obtained

using ESQUIRE-LC or APEXII FT-ICRMS instruments. Optical rotations were determined with a Perkin-Elmer model 241 polarimeter. The IR spectra were run on a Nicolet Impact 400 grating infrared spectrophotometer. The UV spectra were taken on a Perkin-Elmer-554 spectrometer. The 1D- and 2D-NMR spectra were recorded with a BRUKER IVANCE 500 spectrometer. Chemical shifts (δ) are given with TMS as an internal standard. Silica gel precoated plates (Qingdao Ocean Chemical Co., Qingdao, China) were used in TLC. Detection was carried out by spraying with 10% H₂SO₄ solution followed by heating.

3.2 Plant material

The aerial plants of *L. foenum-graecum* were collected in Kunming City, Yunnan Province of China, in June 2001, and were identified by Prof. Zhang Ji. A voucher specimen (No. 0108127) is deposited in the Institute of Chinese Materia Medica, National Institute for the Control of Pharmaceutical and Biological Products.

3.3 Extraction and isolation

The aerial parts of *L. foenum-graecum* (3.0 kg) were extracted with 70% EtOH $(2 \times 52 \text{ l})$ under reflux. The combined filtrate was divided into petroleum ether, CH₂Cl₂, and remaining ethanol fractions. The CH₂Cl₂ fraction (45 g) was purified by repeated silica gel column chromatography (ϕ 9 × 120 cm) eluted with a CHCl₃—MeOH (85:15) gradient system and ODS HPLC (MeOH—H₂O, 9:1) to give compound **2** (6 mg). The remaining ethanol fractions (160 g) were absorbed on a Diaion SP825 column, and then sequentially eluted with H₂O and EtOH. The fraction eluted with 50% EtOH (25 g) was subjected

to silica gel column chromatography (ϕ 7 × 100 cm) using CHCl₃ —MeOH gradient system to yield fractions I–VIII. Fraction II was purified by repeated silica gel column chromatography and a reversed-phase column (Rp18, ϕ 3.5 × 60 cm) using 65% MeOH as eluent to give compound **1** (7 mg).

3.3.1 Lysimachiagenoside A (1)

White powder, $[\alpha]_D^{24} - 3.6 (c = 0.14, MeOH);$ UV (MeOH) λ_{max} (nm) (log ε): 205 (4.00); IR (KBr) ν_{max} (cm⁻¹): 3441 (OH) and 1239 (C=C); ESI-MS *m*/*z*: 1115.6 [M+Na]⁺; HR-FAB-MS *m*/*z*: 1115.5574 [M+Na]⁺ (calcd for C₅₃H₈₈O₂₃Na, 1115.5614).

3.3.2 21-O-angeloylbarringtogenol C (2)

White powder, $[\alpha]_D^{24} - 21.6$ (c = 0.14, MeOH); UV (MeOH) λ_{max} (nm) (log ε): 204 (4.14); IR (KBr) ν_{max} (cm⁻¹): 3430 (OH), 1715 (C=O) and 1254 (C=C); ESI-MS *m/z*: 595 [M+Na]⁺; HR-FAB-MS *m/z*: 595.3979 [M+Na]⁺ (calcd for C₃₅H₅₆O₆Na, 595.3976).

References

- The Health Administration of Beijing, *Beijing Standard of Traditional Chinese Medicine* (in Chinese) (Capital Normal University Press, Beijing, 1998), p. 274.
- [2] I. Kitagawa, I. Yosioka, and A.S. Matasuda, *Chem. Pharm. Bull.* **20**, 2226 (1972).
- [3] H. Kohda, O. Takeda, and S.M. Tanaka, *Chem. Pharm. Bull.* 37, 3304 (1989).
- [4] H. Kitagara, O. Takeda, S.M. Tanaka, and I.C. Yosika, *Chem. Pharm. Bull.* 15, 1435 (1967).
- [5] K. Yasukawa and M.F. Takido, *Planta Med.* 59, 578 (1993).
- [6] K. Yasukawa, H. Ogawa, and M.T. Takido, *Phytochemistry* 29, 1707 (1990).
- [7] K. Takao and K.H. Lee, J. Nat. Prod. 49, 650 (1986).